Vocabulary

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Study the vocabulary words and definitions below.

acid	any of a group of compounds that produce positively charged hydrogen (H+) ions when dissolved in water
base	any of a group of compounds that produce negatively charged hydroxide (OH) ⁻ ions when dissolved in water
concentration	the amount of solute per unit of solution <i>Example</i> : If a beaker of sugar water has half of its volume made of sugar, then it has a 50% concentration of sugar by volume.
dilute	to decrease the amount of solute as compared to the amount of solvent in a solution <i>Example</i> : To dilute sugar water, add water to the solution.
indicator	a chemical that is one color when in the presence of acids and is a different color when in the presence of bases
ion	a charged particle, atom, or molecule

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litmus paper a type of paper used to indicate the presence of acids or bases

neutralization the reaction between a base and an acid which produces water and a salt

neutral solution a solution that is neither an acid nor a base

phenolphthalein a liquid indicator used to show the presence of bases (pronunciation: fee-nol'-thal-e-un)

salt any of a group of compounds

distinguished by being formed from a metal and nonmetal that are ionically

bonded

Examples include:

NaCl (sodium chloride, table salt); MgCl (magnesium chloride, Epsom salt); and NaF (sodium fluoride, the active ingredient in many toothpastes)